

"Science is the disciplined way we turn curious questions into repeatable answers about how the world actually works — and sometimes those answers are urgent."

Short answer: 212 $\mu\text{g/L}$ (micrograms per liter, which equals 212 parts-per-billion) of benzene in water is high and would generally be considered a serious contamination for drinking water. It greatly exceeds commonly used health-based limits and should prompt immediate caution and professional follow-up.

Why that's serious — key points

- Units and magnitude: 212 $\mu\text{g/L}$ = 212 ppb. Drinking-water standards are usually in single digits to low tens of $\mu\text{g/L}$, so 212 $\mu\text{g/L}$ is well above normal regulatory limits.
- Regulatory benchmarks:
 - U.S. EPA Maximum Contaminant Level (MCL) for benzene in drinking water = 5 $\mu\text{g/L}$. 212 $\mu\text{g/L}$ is ≈ 42 times that value.
 - World Health Organization guideline value for benzene in drinking water = 10 $\mu\text{g/L}$ (commonly referenced). 212 $\mu\text{g/L}$ is ≈ 21 times that.
- Toxicology: Benzene is an established human carcinogen (IARC Group 1). Long-term ingestion increases risk of blood disorders (bone marrow suppression, aplastic anemia) and leukemia (especially acute myeloid leukemia). Short-term very high exposures affect the central nervous system (dizziness, headache, unconsciousness), but typical concern with contaminated water is chronic exposure and cancer risk.
- Typical background levels in unpolluted drinking water are generally well below 1–10 $\mu\text{g/L}$; 212 $\mu\text{g/L}$ indicates a significant contamination source (fuel, industrial discharge, leaking storage tank, etc.).

What to do now (practical actions)

1. Don't drink or cook with the water. Use bottled water or an alternative safe supply for drinking and food preparation until the issue is resolved.
2. Avoid boiling (boiling can volatilize benzene and increase inhalation exposure in enclosed spaces).
3. Contact your local health department or water utility immediately. They can recommend confirmatory sampling and public-health actions.
4. Arrange confirmatory laboratory testing (GC–MS or equivalent) and sampling of nearby wells or distribution points to define the extent of contamination.
5. Remediation and treatment options include granular activated carbon (GAC) systems, air stripping, and source cleanup — but these should be designed and installed by professionals.
6. Consider medical/public-health consultation if people in the household have had prolonged exposure; public-health authorities can advise on medical screening and risk communication.

Estimating risk (how you would quantify it)

- To estimate lifetime cancer risk you convert concentration to intake (mg/kg-day) using consumption and body weight, then multiply by an oral cancer slope factor from regulatory toxicology sources. With typical drinking assumptions (e.g., 2 L/day, 70 kg adult), 212 $\mu\text{g/L}$ corresponds to an intake on the order of 0.006 mg/kg-day. Using commonly cited slope factors yields lifetime excess cancer risks often on the order of 10^{-4} to 10^{-3} (i.e., roughly one in ten thousand to one in a thousand) for chronic consumption — a level epidemiologists and regulators view as meaningful and actionable. Exact numbers depend on the slope factor and individual intake/body weight.

If you want, I can:

- Compute a specific intake and lifetime risk for a given body weight and daily water consumption, using a chosen slope factor; or
- Help draft questions to ask your local health department, or a checklist for confirmatory sampling and remediation steps.